Applicant: Todd E. Bofinger et al. Attorney's Docket No.: 08935-301001 / M-5064

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## Amendments to the Claims:

This listing of claims replaces all prior versions and listings of claims in the application:

<u>Listing of Claims</u>:

1. (Currently Amended) A method of making a lithiated manganese dioxide for a primary lithium battery comprising:

contacting a manganese dioxide with a lithium ion source at a lithiation temperature sufficient to substantially between 40° and 120°C to replace at least 75% of protons in the manganese dioxide with lithium ions; and

heating the manganese dioxide at a water removal temperature sufficient to substantially remove residual and surface water to produce a lithiated manganese dioxide having an X-ray diffraction pattern substantially similar to the X-ray diffraction pattern of the manganese dioxide prior to lithiation.

- 2. (Original) The method of claim 1, wherein the manganese dioxide is persulfate derived chemical manganese dioxide.
- 3. (Original) The method of claim 1, wherein the manganese dioxide is gammamanganese dioxide.
- 4. (Original) The method of claim 1, wherein the lithium ion source is an aqueous solution including a lithium salt.
  - 5. (Original) The method of claim 4, wherein the lithium salt is a lithium hydroxide.
- 6. (Currently Amended) The method of claim 1, wherein the lithiation temperature is between 40 C 60° and 100°C.
- 7. (Currently Amended) The method of claim 1, wherein the water removal temperature is between 180°C and 500°C.

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8. (Currently Amended) The method of claim 1, wherein the water removal temperature is between 200°C and 460°C.

9. (Currently Amended) A method of making a cathode for a battery comprising: contacting a manganese dioxide with a lithium ion source;

heating the manganese dioxide to a temperature between 40°C and 120°C to produce a lithiated manganese dioxide in which at least 75% of protons in the manganese dioxide are replaced with lithium ions and having an X-ray diffraction pattern substantially similar to the X-ray diffraction pattern of the manganese dioxide prior to lithiation; and

coating a current collector with a composition including a carbon source, and the cathode active material, wherein the cathode active material includes a manganese dioxide.

- 10. (Original) The method of claim 9, wherein the manganese dioxide is persulfate derived chemical manganese dioxide.
- 11. (Original) The method of claim 9, wherein the manganese dioxide is gamma-manganese dioxide.
- 12. (Original) The method of claim 9, wherein the lithium ion source is an aqueous solution including a lithium salt.
  - 13. (Original) The method of claim 12, wherein the lithium salt is a lithium hydroxide.
- 14. (Currently Amended) The method of claim 9, wherein the lithiation temperature is between [[4]]60°C and 100°C.
- 15. (Currently Amended) The method of claim 9, wherein the water removal temperature is between 180°C and 500°C.

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16. (Currently Amended) The method of claim 9, wherein the water removal temperature is between 200°C and 460°C.

17-21. (Cancelled).